

The Ti-Si-C System (Titanium-Silicon-Carbon)

Debashis Bandyopadhyay

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In the present article different isothermals of Ti-Si-C system at temperatures ranging from 1250 to 2877 °C, previously reported by [1966Bru], [1989Tou], [1991Wak], and [1993Sei], were assessed and redrawn on the basis of the recently reported binary alloy phase diagram of Ti-Si, Ti-C, and Si-C.

1. Ti-C System

The assessed phase diagram of Ti-C system shown in Fig. 1 is taken from [1998Oka]. Other recent work done on this system has been published by [1993Oht], [1995Alb], [1996Sei1], and [1996Jon]. The system consists of two terminal solids α -Ti and β -Ti, a refractory monocarbide TiC. Other phases present are liquid and graphite (C). Two eutectic reaction and one peritectoid reaction appear in this system at 1646, 2776, and 920 °C, respectively. Crystal structure data shown Table 1 are taken from [Massalski1].

Debashis Bandyopadhyay, Birla Institute of Technology and Science, Pilani, Rajasthan-333031, India. Contact e-mail: bandy@bits-pilani.ac.in.

2. Ti-Si System

The equilibrium phase diagram of Ti-Si binary system shown in Fig. 2 is taken from [1996Sei2]. An important contribution is from [1989Vah]. The Ti-Si system is characterized by three eutectic and two peritectic reactions. The different intermediate phases present in this system are Ti_3Si , Ti_5Si_3 , Ti_5Si_4 , Ti_6Si_5 , $TiSi$, and $TiSi_2$. Crystal structure data shown in Table 2 are taken from [Massalski1].

3. Si-C System

Very few experimental data are available on Si-C binary system. The assessed phase diagram in Fig. 3 is taken from [1992Luk]. This system includes one peritectic reaction and

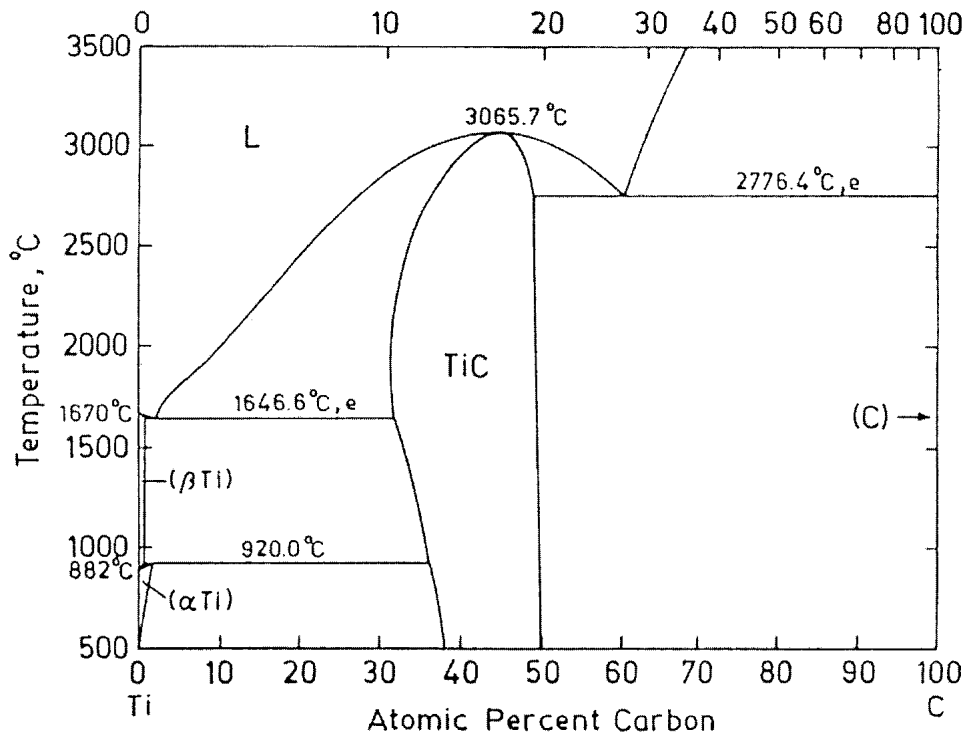


Fig. 1 Ti-C phase diagram

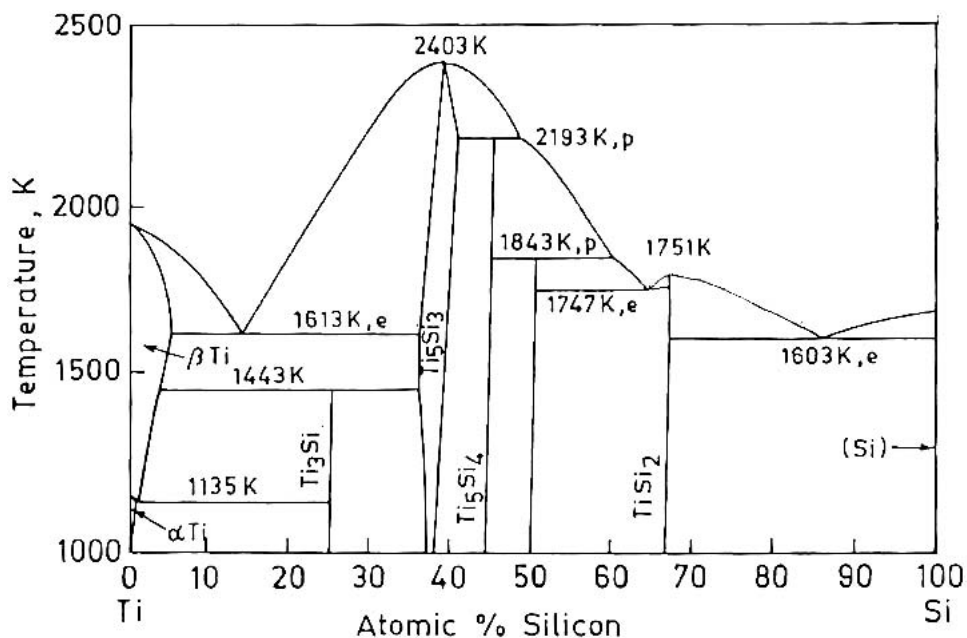


Fig. 2 Ti-Si phase diagram

Table 1 Ti-C crystal structure data

Phase	Composition, at. % C	Pearson symbol	Space group	Strukturbericht designation	Prototype
(β -Ti)	0 to 0.6	<i>cI2</i>	<i>Im</i> $\bar{3}m$	A2	W
(α -Ti)	0 to 1.6	<i>hP2</i>	<i>P6</i> $\bar{3}/mmc$	A2	Mg
γ -TiC	~32 to 48.8	<i>cF8</i>	<i>Fm</i> $\bar{3}m$	B1	NaCl
Ti ₂ C	~32 to 36	<i>cF48</i>	<i>Fm</i> $\bar{3}m$	B1	NaCl

Table 2 Ti-Si crystal structure data

Phase	Composition, at.% Si	Pearson symbol	Space group	Strukturbericht designation	Prototype
(α -Ti)	0 to 0.5	<i>hP2</i>	<i>P6</i> $\bar{3}/mmc$	A3	Mg
(β -Ti)	0 to 3.5	<i>cI2</i>	<i>Im</i> $\bar{3}m$	A2	W
Ti ₃ Si	25	<i>tP32</i>	<i>P4</i> $\bar{2}/n$...	Ti ₃ P
Ti ₅ Si ₃	35.5 to 39.5	<i>hP16</i>	<i>P6</i> $\bar{3}/mcm$	<i>D8</i> ₈	Mn ₅ Si ₃
Ti ₅ Si ₄	44.4	<i>tP36</i>	<i>P4</i> ₁ 2 ₁ 2	...	Zr ₅ Si ₄
Ti ₆ Si ₅	45.5	(a)
TiSi	50	<i>oP8</i>	<i>Pmm</i> 2	...	TiSi
		<i>oP8</i>	<i>Pnma</i>	B27	FeB
		<i>oP8</i>	<i>Pnma</i>	B27	FeB
TiSi ₂	66.7	<i>oF24</i>	<i>Fddd</i>	C54	TiSi ₂
(Si)	100	<i>cF8</i>	<i>Fd</i> $\bar{3}m$	A4	C(diamond)

(a) Tetragonal, related to σ (*D8*₈)

one eutectic reaction at 2545 and 1404 °C, respectively. Crystal structure data shown in Table 3 are taken from [Massalski1].

4. Ti-Si-C System

Phase stability in the Ti-Si-C ternary system was investigated by [1992Sam]. They have constructed an isothermal

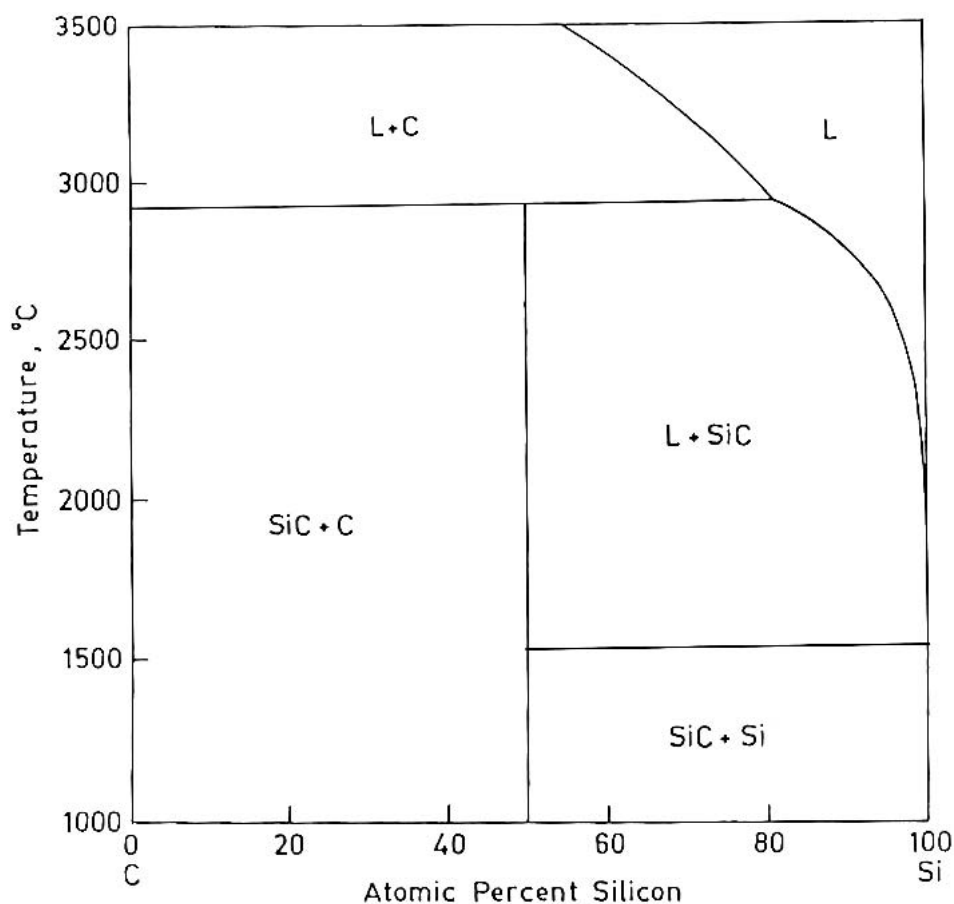


Fig. 3 Si-C phase diagram

Table 3 Si-C crystal structure data

Phase	Composition at.% Si	Pearson symbol	Space group	Strukturbericht designation	Prototype
(Si)	0	<i>cF8</i>	$F\bar{4}3m$	A4	C(diamond)
SiC or β -SiC	50	<i>cF8</i>	$F\bar{4}3m$	B3	ZnS
C or graphite	100	<i>hP4</i>	$P6_3/mmc$	A9	C(graphite)
α -SiC(a)	50	(b)
Amorphous (c)	39 to 61
High Pressure					
SiC II	...	tI_4	$I4_1/amd$	A5	β -Sn

(a) Other SiC poly-types have been reported. (b) Hexagonal. (c) SiC is a line compound even amorphous. 39 to 61 at.% of SiC could be a mixture of SiC and Si or SiC and C.

section at 1200 °C and 240 MPa pressure. A previous study was done on this system by [1966Bru]. The isothermal section at the above temperature and pressure constructed by [1992Sam] shows the existence of $Ti_3Si(c)$ phase that was not present in the study of [1966Bru]. [1975Ava] investigated the possibility of isolating a single phase using the method of crystallization from the gas phase in this system in the temperature range of 1100-1700 °C. Interaction between titanium and SiC was studied by [1990Cho] by electron microprobe, scanning electron microscopy (SEM), and

transmission electron microscopy (TEM) techniques. [1989Tou] reviewed the Gibbs free energies of formation of the solid compounds in the Ti-Si-C system and calculated an isothermal section at 1227 °C. Extensive work on this system was done by [1993Sei]. On the basis of the work done by [1966Bru], [1989Tou], [1991Wak], and [1993Sei] together with recently assessed binaries, the ternary Ti-Si-C alloy phase diagram has been assessed and from this has been drawn the isothermal sections from 1250 to 2877 °C that are shown in Fig. 4-7.

Section I: Basic and Applied Research

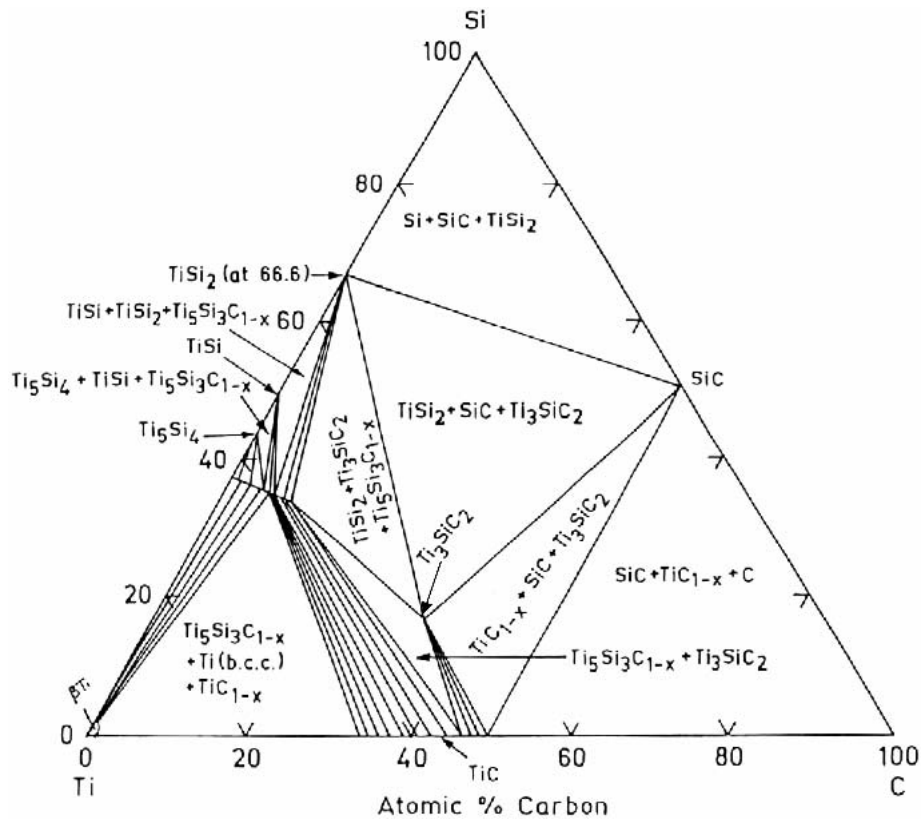


Fig. 4 Isothermal sections of Ti-Si-C system at 1250 °C

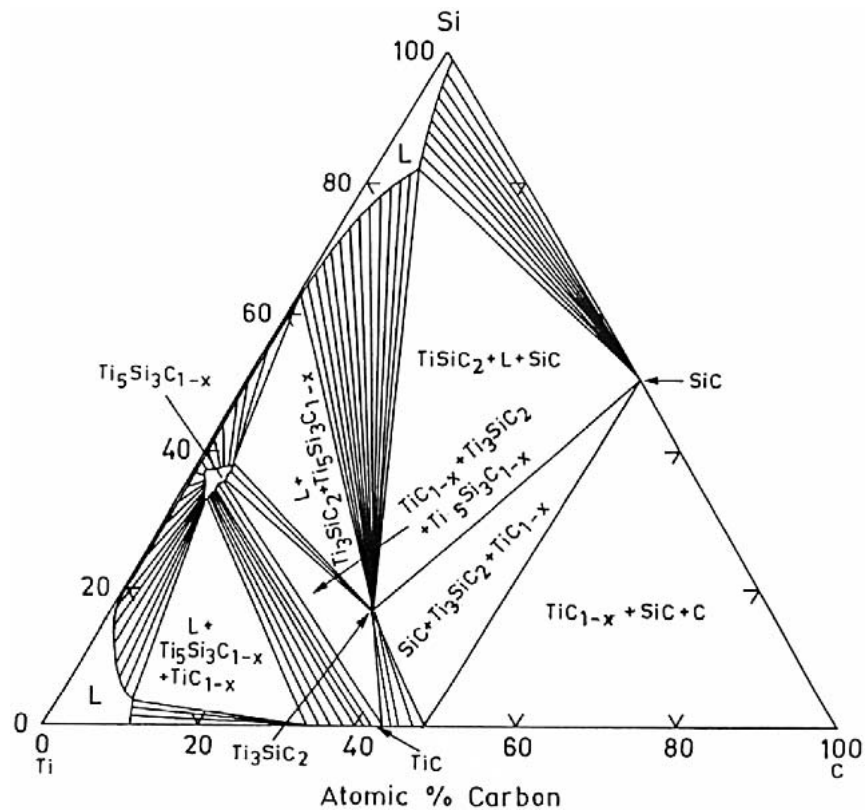


Fig. 5 Isothermal sections of Ti-Si-C system at 2222 °C

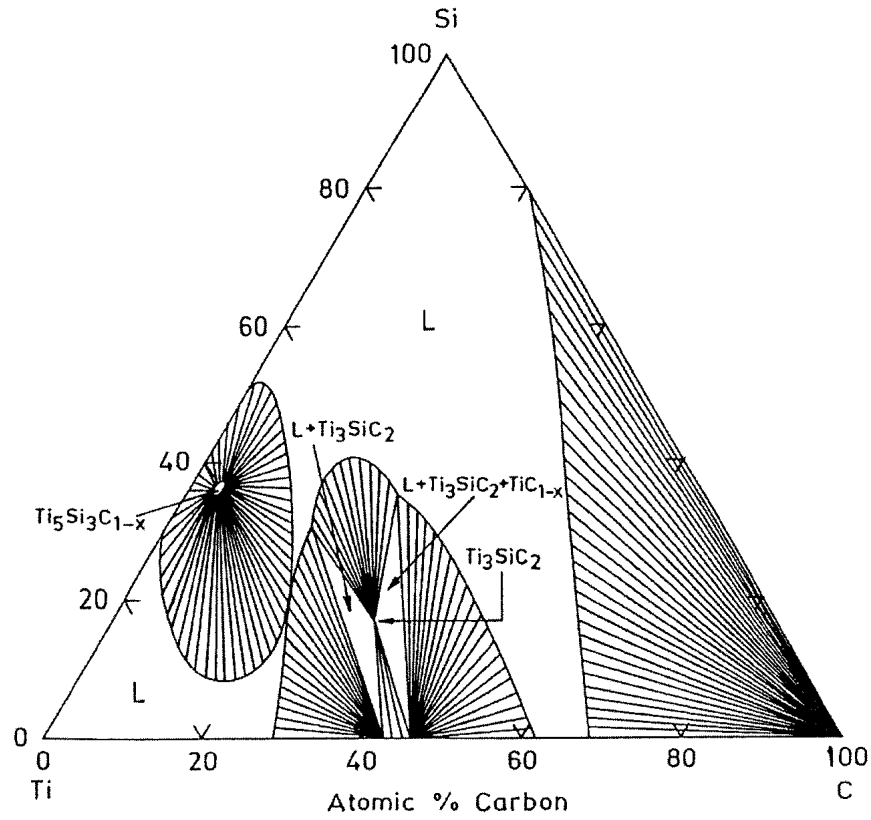


Fig. 6 Isothermal sections of Ti-Si-C system at 2827 °C

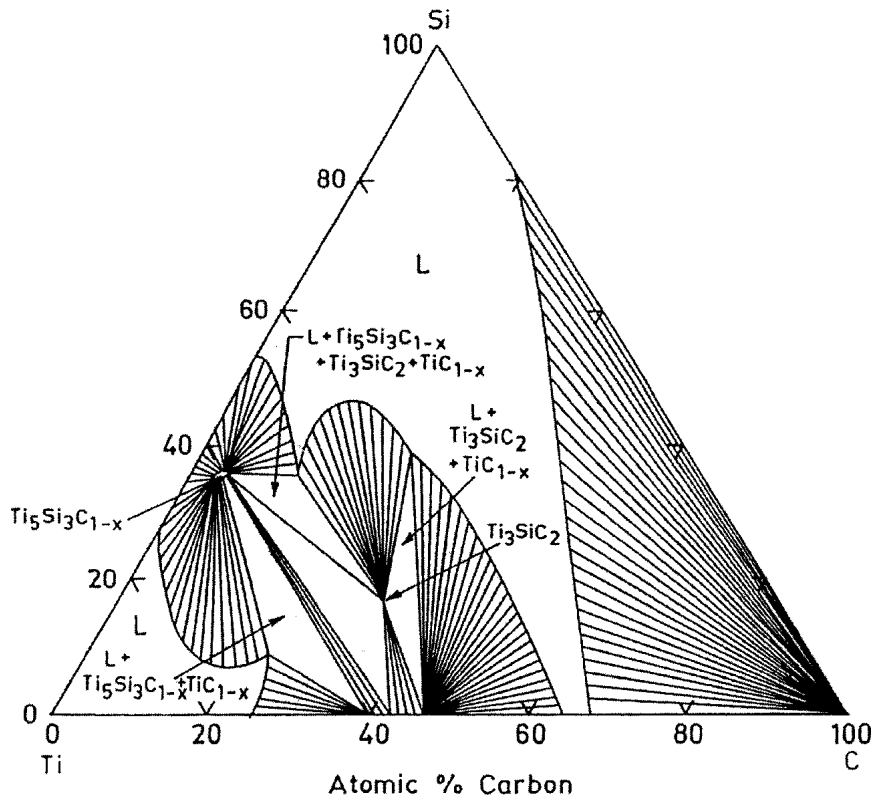


Fig. 7 Isothermal sections of Ti-Si-C system at 2877 °C

Section I: Basic and Applied Research

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